



# Simon Silver

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## Bacterial view of the chemical periodic table: Genes (and proteins) for each element of the chemical periodic table

Microbial activities often provide the basis for useful environmental and agricultural biotechnology, as well as frequently causing problems. Essentially all bacteria have genes for toxic metal ion resistances and these include *thaSeO<sub>2</sub>* for Ag AsO<sub>4</sub><sup>3-</sup>, Cd<sup>2+</sup>, Co<sup>2+</sup>, CrO<sub>4</sub><sup>2-</sup>, Cu<sup>2+</sup>, Hg<sup>2+</sup>, Ni<sup>2+</sup>, Pb<sup>2+</sup>, TeO<sub>3</sub><sup>2-</sup>, Tl<sup>+</sup>, and Zn<sup>2+</sup>. Resistance to inorganic Hg and to organomercurials such as CH<sub>3</sub>Hg<sup>+</sup> and phenylmercury involve a series of metal-binding and membrane transport proteins as well as the enzymes mercuric reductase and organomercurial lyase. Hg is methylated and demethylated by microbial processes. Methylmercury of concern in human food is of microbial origin and microbial bioremediation and phytoremediation can clean polluted sites. Arsenic resistance and metabolizing systems occur in three forms, the widely-distributed *aro* operon that is present in most bacterial genomes and many plasmids, the more recently-recognized the *aso* genes for the periplasmic arsenite oxidase that serves as initial electron donor in aerobic resistance to arsenite and the functionally-related *arr* genes for arsenate reductase that serves as terminal electron acceptor in anaerobic respiration. The largest group of resistance systems function by energy-dependent efflux of toxic ions. Some of the efflux resistance systems are ATPases and others are chemiosmotic ion/proton exchangers. For example Cd<sup>2+</sup> efflux pumps of bacteria are either inner membrane P-type, ATPases or three polypeptide RND chemiosmotic complexes consisting of an inner membrane pump, a periplasmic- bridging protein and an outer membrane channel. Silver compounds are increasingly used in industrial, environmental and medical applications. A cluster of 9 silver-specific genes make proteins that bind extracellular Ag<sup>+</sup> or internalized Ag<sup>+</sup> from the cells, using membrane potential or ATP hydrolysis for energy. The SilE periplasmic Ag<sup>+</sup> binding protein is an unusual small soluble protein that binds 5 Ag<sup>+</sup> ions with 10 histidine residues.

## Biography

Simon Silver received his PhD at the University of Illinois Urbana-Champaign in 1967 under the direction of Dr. James C. Kirby. He did postdoctoral work at the University of California Berkeley with Dr. Donald J. Cane. He joined the faculty at the University of Illinois College of Medicine in 1970 and was promoted to full professor in 1976. He has been a member of the National Academy of Sciences since 1987. He has served as editor-in-chief of the Journal of Clinical Experimental Pathology and as editor of the International Journal of Environmental Research and Public Health. He has also served as editor of the International Journal of Environmental Research and Public Health. He has published over 200 papers and edited 9 published monographs. He was Editor in Chief of 2 journals and editor or editorial board member of more than a dozen more.

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