



J P E F H S B E B U J P O P G) F Y B D I M P S P D Z D M P I F Y B O ' V O H M M F V S P U V T G M P S J E B

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Abstract

Hexachlorocyclohexane (HCH) isomers are reported to persist in the environment long after their usage is discontinued. 3 OHXURWXVDÄ ZKULGD URW IXQJXV z x d á YhAì Q4v1#)h`à_H,i-\$ H SHUVLVWHQW LVRPHUV LQGLYLGXDOO\ DV ZHOO DV D PL[WXUH LQ D 5DSHU1V FRP FRPSOHWHO\ ZKHUHDV RWKHU LVRPHUV GHJUDGHG 7KH GHJUDGDWLRQ UI 7KH P\FHOLXP ELRPDVV ZDV IUHH IURP ++ UHVLGXHV EXW DFFXPXODWHG DER Presence of intermediate metabolites was not detected indicating complete mineralization of HCH isomers. Ability of 3 À R U L G D to degrade HCH isomers was further studied in soil by amendment with spent mushroom substrate (SMS). 606 DGGLWLRQ FRXOG PDUJLQDOO\ LQFUHDVH GHJUDGDWLRQ RI . DQG / ++ RI ++ :KHQ WKH VWXG\ ZDV UHSHDWHG VLPLODU WUHQG ZDV REVHUYHG 7KH 439-570 days in un-amended soil while 37-42 days in SMS amended soil. For other 3 stable isomers the half-life was reduced from 686-828 to 88-125 days by SMS amendments. These results indicate that SMS from 3 À R U L G D cultivation can be utilized for bioremediation of HCH contaminated site.

Keywords Degradation; Hexachlorocyclohexane; Metabolites; Pleurotus orida; Spent mushroom substrate

Abbreviations HCH: Hexachlorocyclohexane; LOD: Limit of Detection; P: Pleurotus

Introduction

Researcher's name is 7210735. Multiple instances of the word "Introduction" are present in the text, suggesting a repetitive or corrupted section. The text also includes a DOI: 10.4172/2155-6199.1000280.

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(purity 99.9%), 1,2,4,5-tetrachlorobenzene (purity 99.4%), 1,2,3,5-tetrachlorobenzene (purity 97.5%), 1,2,3,4-tetrachlorobenzene (purity 99.0%), and pentachlorobenzene (purity 99.8%) were procured from Sigma-Aldrich, Germany. All chemicals and reagents used were of analytical grade.

Standard solution

Stock solutions (1000 µg mL⁻¹) of α -, β -, γ -, δ -HCH and all the metabolites were prepared in n-hexane. Working standards were

mode (mass acquisition range m/z 40-450) using ionization energy of 70 eV. The gas chromatograph was operated in split less mode with injector temperature of 250°C and ion source temperature of 200°C. A capillary column, Agilent DB-5ms (30 m × 0.25 mm i.d. 0.25 μm; stationary phase, 5% phenyl 95% methyl polysiloxane) was used for analy-



had degraded whether treated individually or as a mixture with other isomers. The mycelium biomass from γ -HCH treated medium was also that of *P. orida* not only degraded α - and γ -HCH, it degraded the more free from its residues. γ -HCH, which is known to be highly stable in persistent α - and γ -isomers equally fast. The analysis of extracts from the environment, also degraded very fast, with about 91% being lost from medium/mycelium did not show any additional peak corresponding from the treated medium after 20 days. A small amount (3.23%) of any of the metabolites mentioned above. This was also confirmed γ -HCH was found adsorbed by the mycelium. However degradation of α -HCH was not as fast as the other 2 isomers. Paper's complete medium is given in Figure 1.

Degradation of HCH isomers in soil

Degradation of HCH isomers was slow in soil without SMS compared to soil with SMS. It was also observed that there was a difference in the rate of degradation of all isomers in the presence of SMS (Table 3). In the initial 30 days period there was no significant difference in the rate of degradation of all isomers. But after 40 days period about 12-17% of α -, β - and γ -HCH was lost from soil while about 38% γ -HCH was lost. After a period of 60 days about 61-67% γ -HCH was lost from soil with SMS compared to about 20-30% loss of the other isomers. In soil without SMS the loss of HCH residues were about 4-9% in 60 days. The experiment was repeated and a similar trend was observed (Table 4). Degradation of HCH isomers from soil is given in Figure 2.

Dissipation kinetics of HCH isomers in soil

Residue dissipation of HCH isomers was studied by subjecting the data to first order kinetics; $C = C_0 e^{-kt}$, where C is the concentration at time t ; C_0 is the initial concentration, k is the rate constant for dissipation and $t_{1/2} = \ln(2)/k$. It was observed that the dissipation of all HCH isomers followed first order rate kinetics. In soil without SMS the half-life of degradation of α -, β - and γ -HCH was in the range of 748-828 days. But when SMS was added to the soil it was substantially reduced.

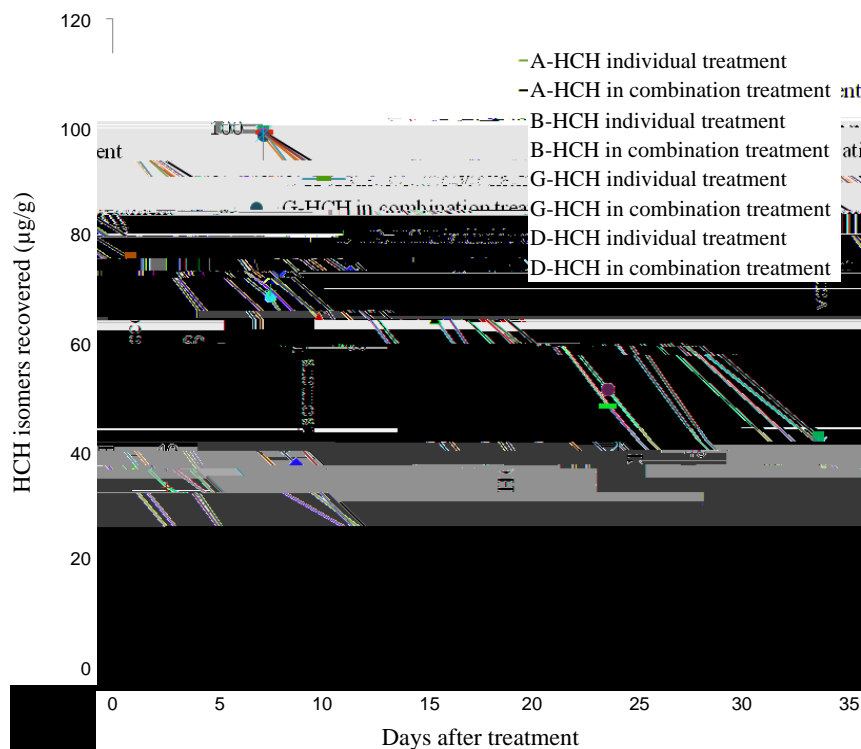


Figure 1: α -, β -, γ - and δ -HCH degradation of HCH isomers individually and as mixture of 4 isomers.

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