

Editorial on Plant-Mediated Synthesis of Iron Nanoparticles

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Abstract

Plant-mediated synthesis of iron nanoparticles is a promising approach to reduce the metal ions in a shorter time as compared to microbes. Depending upon plant type and concentration of phytochemicals, nanoparticles are synthesized within a few minutes or hours. The synthesis of nanoparticles is the obligatory constraint of aseptic conditions, which requires along with the easy availability of plants in nature, make them more preferred biological resources than microbes. Magnetic nanoparticles have emerged as a new class of important nanoparticles as they possess many exceptional properties like superparamagnetism, high coercivity, and so forth. These nanoparticles, when synthesized by conventional methods, have several limitations such as the following:

(b) The magnetic nanoparticles synthesized by conventional methods cannot be used in biomedical applications wherein nonpolar organic solvents are used.

A review paper on microbial synthesis of magnetic nanoparticles has been published by Abhilash et al. Two principal mechanisms, namely, biologically induced mineralization (BIM) and biologically controlled mineralization (BCM), have been discussed, by which microbes synthesize iron oxide nanoparticles. However, the synthesis of iron oxide nanoparticles has remained a relatively unexplored research area. To the best of our knowledge, a comprehensive review on the synthesis of iron nanoparticles by plant resources and provide an updated consolidation of the published literature regarding biosynthesis of magnetic nanoparticles by plant resources along with its advantages and future scope of work in this area. The paper is divided into three main sections depending upon the usage of plant resources for nanoparticle synthesis, namely, in the form of the extract, whole plant part (biomass), and as a template.

Conventional nanoparticle synthesis methods like attrition and pyrolysis have drawbacks such as defective surface formation, low production rate, high cost of manufacturing, and large energy requirement. Chemical synthesis methods (e.g., chemical reduction, sol gel technique, etc.) involve the usage of toxic chemicals, formation of hazardous byproducts, and contamination from precursor chemicals. Hence, there is a need for eco-friendly procedures for nanoparticle synthesis. Some of the distinct advantages that biological synthesis protocols have over the conventionally used physical and chemical methods are:

(a) The active biological component like enzyme itself acts as a reducing and capping agent, thereby reducing the overall cost of the synthesis process.

A very wide range of biological resources like microorganisms (bacteria, yeast, fungi, algae, and viruses) and plants can be used for the synthesis of iron nanoparticles. However, the synthesis of iron nanoparticles has gained importance only in the recent years. Plant extracts

reduce the metal ions in a shorter time as compared to microbes. Depending upon plant type and concentration of phytochemicals, nanoparticles are synthesized within a few minutes or hours. The synthesis of nanoparticles is the obligatory constraint of aseptic conditions, which requires along with the easy availability of plants in nature, make them more preferred biological resources than microbes. Magnetic nanoparticles have emerged as a new class of important nanoparticles as they possess many exceptional properties like superparamagnetism, high coercivity, and so forth. These nanoparticles, when synthesized by conventional methods, have several limitations such as the following:

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