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Editorial on Plant-Mediated Synthesis of Iron Nanoparticles

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Abstract

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Conventional nanoparticle synthesis methods like attrition and pyrolysis have drawbacks such as defective surface formation, low production rate, high cost of manufacturing, and large energy requirement. Chemical synthesis methods (e.g., chemical reduction, sol gel technique, etc.) involve the usage of toxic chemicals, formation of hazardous byproducts, and contamination from precursor chemicals Hence, there is c" i tq ykp i"pggf" vq" fgxgnqr" engcp." pqpvqzke." cpf" gpxktqp o gpv/ friendly procedures for nanoparticle synthesis. Some of the distinct advantages that biological synthesis protocols have over the conventionally used physical and chemical methods are

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(b) the active biological component like enzyme itself acts as a reducing and capping agent, thereby reducing the overall cost of the synthesis process

A very wide range of biological resources like microorganisms (bacteria, yeast, fungi, algae, and viruses) and plants can be used hqt"pcpqrctvkeng"u{pvjguku0" Y jkng" o ketqdg/dcugf"rtqvqeqnu"jcxg" dggp" fgxgnqrgf"htqo"vjg"ew o wncvkxg"tgugctej "g qtvu"qh"ugxgtcn" cwvjqtu." rncpv/ o gfkcvgf" dkqnqikecn" u{pvjguku" qh" pcpqrctvkengu" has gained importance only in the recent years . Plant extracts

reduce the metal ions in a shorter time as compared to microbes. Depending upon plant type and concentration of phytochemicals, nanoparticles are synthesized within a few minutes or hours, y jgtgcu" oketqqticpku o/dcugf" ogvjqfu" tgswktg" c" nqpigt" vk og0" Vjg" oclqt"ftcydcem"qh" oketqdg/ ogfkcvgf"pcpqrctvkeng"u {pvjguku" is the obligatory constraint of aseptic conditions, which requires vtckpgf" uvc ." cpf" tckugu" vjg" uecnkpi/wr" equv0" Cnn" vjgug" tgcuqpu." along with the easy availability of plants in nature, make them more preferred biological resources than microbes.

Magnetic nanoparticles have emerged as a new class of important nanoparticles as they possess many exceptional properties like superparamagnetism, high coercivity, and so forth. These nanoparticles, when synthesized by conventional methods, have several limitations such as the following.

*c+" $Pcpquecng"|gtq/xcngpv"ktqp"rctvkengu"*p \ XK+"u{pvjguk|gf"} by physical and chemical methods are highly reactive in nature and tend to form aggregates, which ultimately results in loss of reactivity.$

(b) The magnetic nanoparticles synthesized by conventional methods cannot be used in biomedical applications wherein nonpolar organic solvents are used .

A review paper on microbial synthesis of magnetic nanoparticles has been published by Abhilash et al. Two principal mechanisms, namely, biologically induced mineralization (BIM) and biologically controlled mineralization (BCM), have been discussed, by which microbes synthesize iron oxide pcpqrctvkengu0" Jqygxgt." rncpv/ogfkcvgf"u{pvjguku"qh" ocipgvke" nanoparticles has remained a relatively unexplored research ctgc" ykvj" vjg" o clqtkv{" qh" rcrgtu" dgkpi" rwdnkujgf" qpn {" kp" vjg" last two years. To the best of our knowledge, a comprehensive tgxkgy"ctvkeng"uwooctk|kpi"vjg"pqvcdng"Lpfkpiu"qh"tgugctejgtu" kp"vjku"Łgnf" jcu"pqv"dggp" rtqfwegf0"Vjku" rcrgt"ckou"vq"Łnn"vjku" lacuna and provide an updated consolidation of the published literature regarding biosynthesis of magnetic nanoparticles by plant resources along with its advantages and future scope of work in this area. The paper is divided into three main sections depending upon the usage of plant resources for nanoparticle synthesis, namely, in the form of the extract, whole plant part (biomass), and as a template.