of the poorly water-soluble drug without any matrix material suspended in dispersion⁴. These can be used to enhance the solubility of drugs that are poorly soluble in water as well as lipid media. As a result of increased solubility, the rate of flooding of the active compound increases and the maximum plasma level is reached faster. This approach is useful for molecules with poor solubility, poor permeability, or both, which poses a significant challenge for the formulators. The reduced particle size renders the possibility of intravenous administration of poorly soluble drugs without any blockade of the blood capillaries. The suspensions can also be lyophilized and into a solid matrix. Apart from these advantages, it also has the advantages of liquid formulations over others 5. In the present review, we are mainly focusing on the different methods of preparation associated merits, demerits, and its pharmaceutical application as drug delivery system.

ADVANTAGES OF NANOSUSPENSION

- Enhance the solubility and bioavailability of drugs
- Suitable for hydrophilic drugs
- Higher drug loading can be achieved
- Dose reduction is possible
- Enhance the physical and chemical stability of drugs
- Provides a passive drug targeting

Preparation of nano-suspension:

The most common approach that has been used for preparing nanosuspensions is micronization by colloid or jet milling ⁶. The method increases the dissolution rate of the drug but does not have any impact on the saturation solubility and thus cannot improve the bioavailability of drugs.

Sucker and co-workers used a precipitation technique to produce nanoparticles by dissolving the drug in a solvent and adding the solvent to a non-solvent that cause precipitation of the fine drug particle ⁷. This has the advantage of using relatively simple and low-cost equipment. However, this created problems in stirring and mixing when taken up for large-scale production. The major challenge of this technique is to avoid crystal growth that occurs on storage due to Ostwald ripening. Mainly there are two methods for preparation of nanosuspensions. The conventional methods of precipitation (Hydrosols²⁵) are called 'Bottom Up technology'. In Bottom Up Technology the drug is dissolved in a solvent, which is then added to nonsolvent to precipitate the crystals. The basic advantage of precipitation technique is the use of simple and low cost equipments. The basic challenge of this technique is that during the precipitation procedure the growing of the drug crystals needs to be controlled by addition of surfactant to avoid formation of microparticles. The limitation of this precipitation technique is that the drug needs to be soluble in into breaking of microparticulate drug to nanosized particles¹², ¹³. AdvaRBBBBK et. al., October - November, 2013, 2(6), 642-649

pressure homogenization, change in crystalline structure of formulation occurs which may be converted to either amorphous or other polymorphic forms²⁶. Alteration in the solid state of the drug particles and the extent of the amorphous portion is determined by X-ray diffraction analysis ²⁹ and supplemented by differential scanning calorimetry analysis²⁵.

Surface Charge (Zeta Potential)

Surface charge properties of the nanosuspensions are studied through zeta potential. The value of particle surface charge indicates the stability of nanosuspensions at the macroscopic level. A minimum zeta potential of ± 30 mV is required for electrostatically stabilized nanosuspensions³⁰, ³¹

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delivery needs for the delivery of peptides and proteins and poorly soluble molecules.

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